

CGA Energy Nexus & Annual Technical Conference 2024

Fuelling the Future

Odorant Systems

Lisa Mazur



Course overview

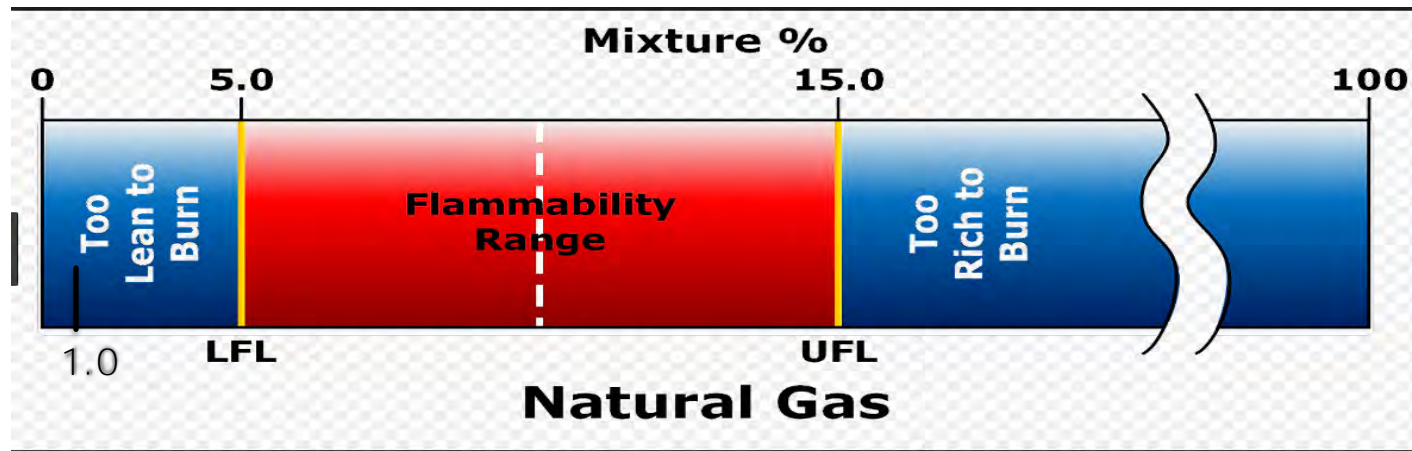
- ▶ Legislation around odorization
- ▶ Characteristics of Mercaptan
- ▶ Various Mercaptan blends
- ▶ Odorant systems and their function
- ▶ Ways to ensure odorization thresholds have been met
- ▶ Constraints/ limitations of use of Mercaptan odorant systems

Industry standards

Natural gas explosive limits

The Lower Explosion Limit (LEL) = 5% Gas in Air

The Upper Explosion Limit (UEL) = 15% Gas in Air



- ▶ The Canadian Standards Association (CSA) Z662 4.21 defines the standards for odorization.
- ▶ Gas is to be readily detectable at concentrations not less than $1/5^{\text{th}}$ of the LEL = 1% Gas in Air (GIA)

Liquid odorant for natural gas

GAS ODORANT CHARACTERISTICS

ISO #13734			
ODORIZING PROPERTIES	CHEMICAL PROPERTIES	PHYSICAL PROPERTIES	TOXICITY PROPERTIES
"Gassy" Unpleasant Persistent Distinctive Not confusable	Stability Non Corrosive Complete combustion	High volatility Low freezing point	Low toxicity
Customers: "...and it must be affordable"			

- ▶ Clear, Potent smelling liquid
- ▶ Mineral oil like consistency
- ▶ 1 Drop will clear this room
- ▶ Very volatile (low vapor pressure)
- ▶ UN Number 3336- Mercaptans, Liquid, Flammable

MERCAPTANS AND SULFIDES BEST MEET THESE PARAMETERS

Mercaptans are also found in Nature (skunks) and daily household items like essential oils

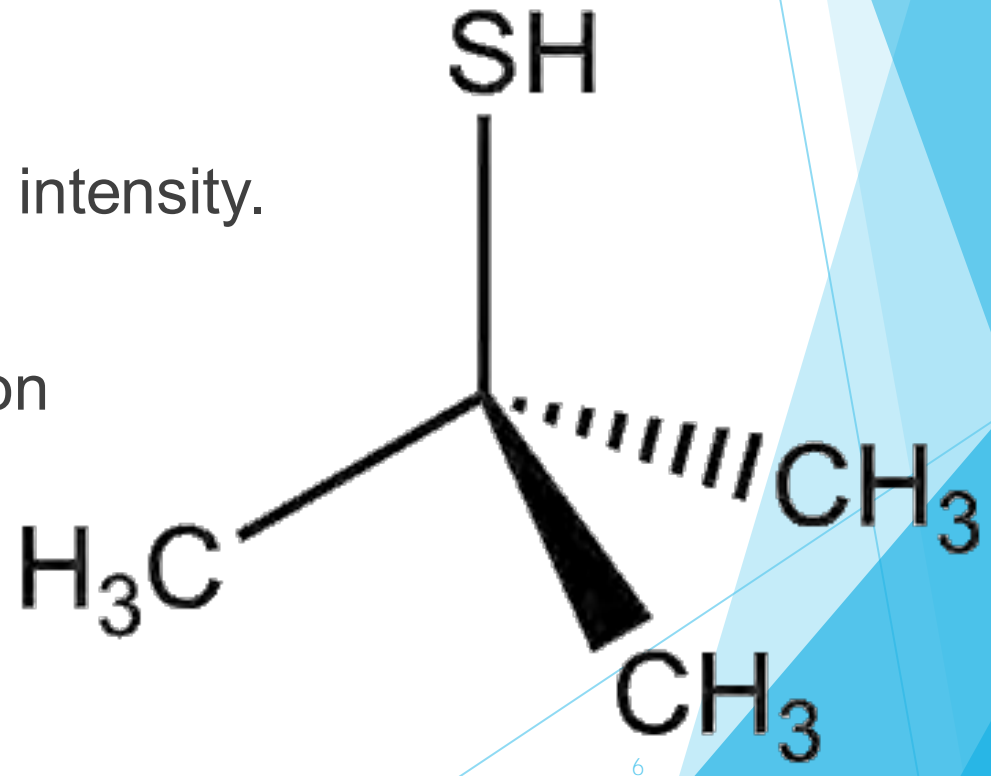
COMMON GAS ODOURANTS BLENDS

Mercaptans					Sulfides			Injection type		Density at 60F (15.5C) in lbs/Gal	Cloud point	
TBM	IPM	NPM	SBM	Ethyl Mercaptan	DMS	MES	THT	Vapor	liquid			
79%	18%	3%								6.76	≤ -50°F -46°C	
80%						20%				6.79		
75%					25%					6.80		
10%	70%	10%			10%					6.87		
50%						50%				6.88		
80%					20%					6.80		
50%							50%			7.46		
30%							70%			7.84		
							100%			8.36	≤-20°F [-29°C]	
				> 99%						7.05		
1 °C 34 °F	-130 °C -202 °F	-113 °C -171 °F	-140 °C -220 °F	-148 °C -234 °F	-98 °C -144 °F	-106 °C -159 °F	-96 °C -141 °F					

Mercaptans

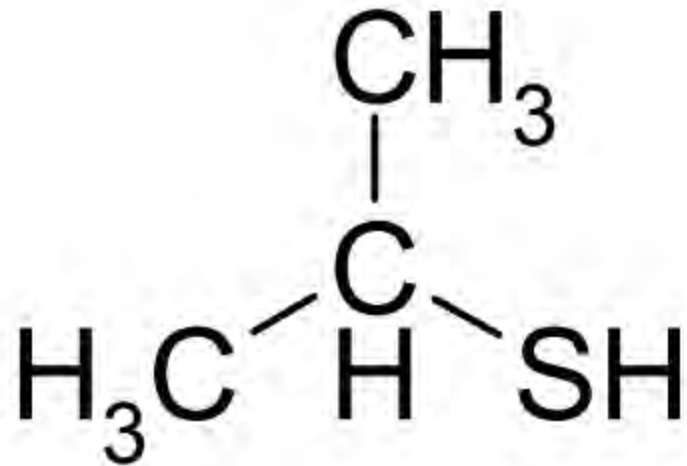
Tertiary Butyl Mercaptan (TBM)

- ▶ Typical gassy odor with very high intensity.
(THE STINK)
- ▶ The highest resistance to oxidation
- ▶ High freezing point (1°C, 34°F)
- ▶ Easily masked and faded



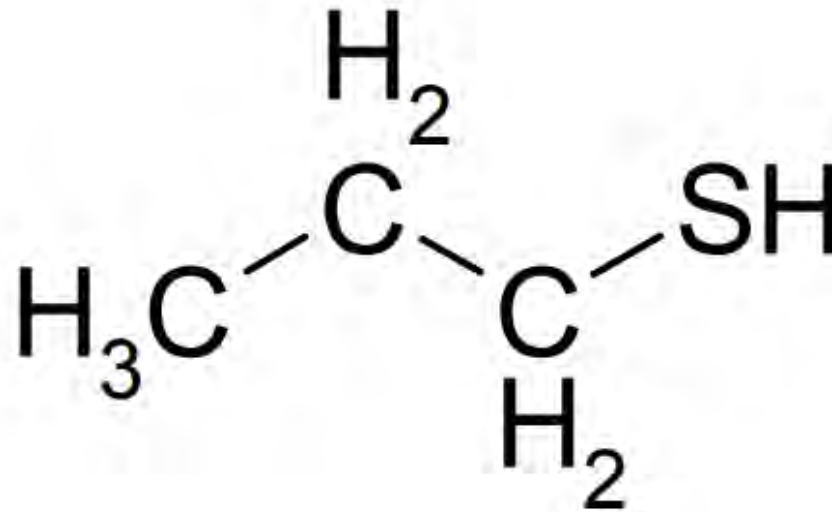
Mercaptans

Iso-Propyl Mercaptan (IPM)



- ▶ 2nd most resistant to oxidation
- ▶ Low freezing point (-130°C, -202°F)

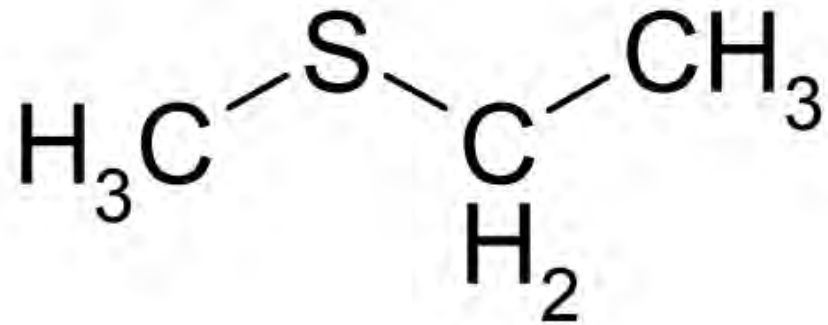
N-Propyl Mercaptan (NPM)



- ▶ Easily oxidized
- ▶ Low freezing point (-113°C, -171°F)

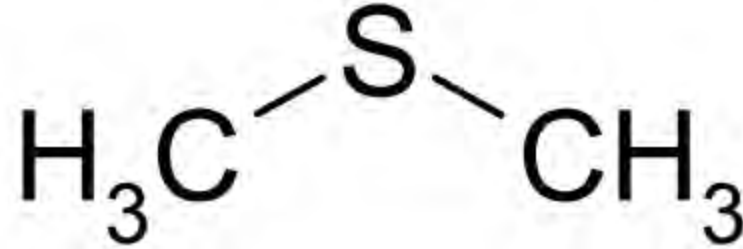
Sulfides

Methyl Ethyl Sulfide (MES)



- ▶ Low freezing point (-106°C, -159°F)
- ▶ Won't oxidize in the presence of rust
- ▶ Good gassy odor
- ▶ Suitable for injection or vaporization type odorizers

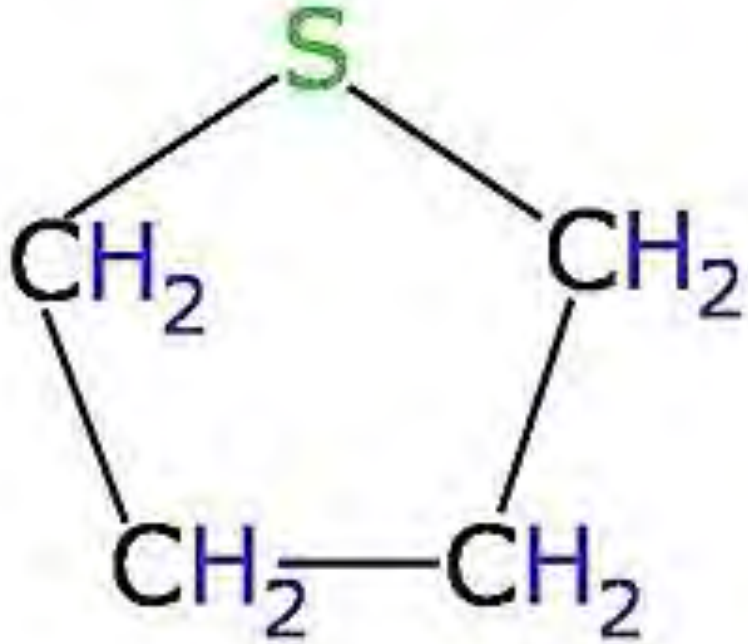
Dimethyl Sulfide (DMS)



- ▶ Low freezing point (-98°C, -144°F)
- ▶ Won't oxidize in the presence of rust
- ▶ No gassy odor

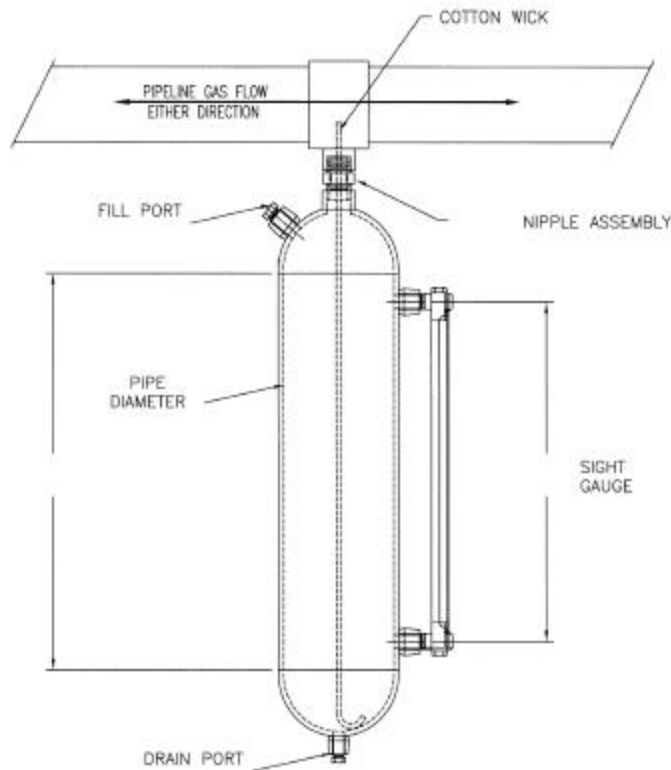
Sulfides

Tetrahydrothiophene (THT)



- ▶ Gassy odor
- ▶ Resistant to pipeline & soil oxidation
- ▶ Used in pure form or as part of TBM blend

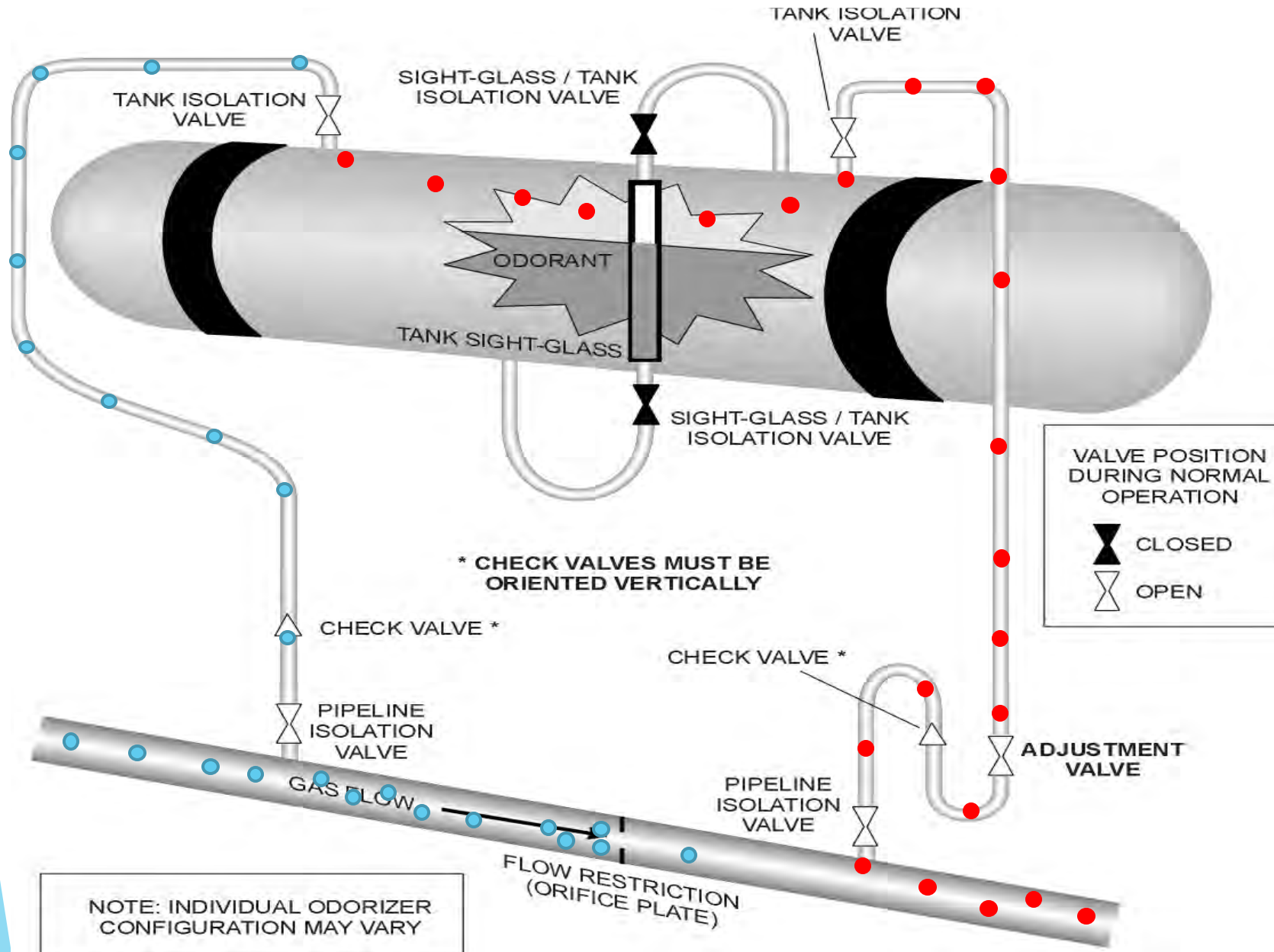
How to we add odorant to our system?



Wick odorizers



Bypass/mini-bypass odorizer



Orifice plate creates
pressure differential

High pressure upstream

Low pressure downstream

Causes gas to find path of least resistance into and out of the odorizer.

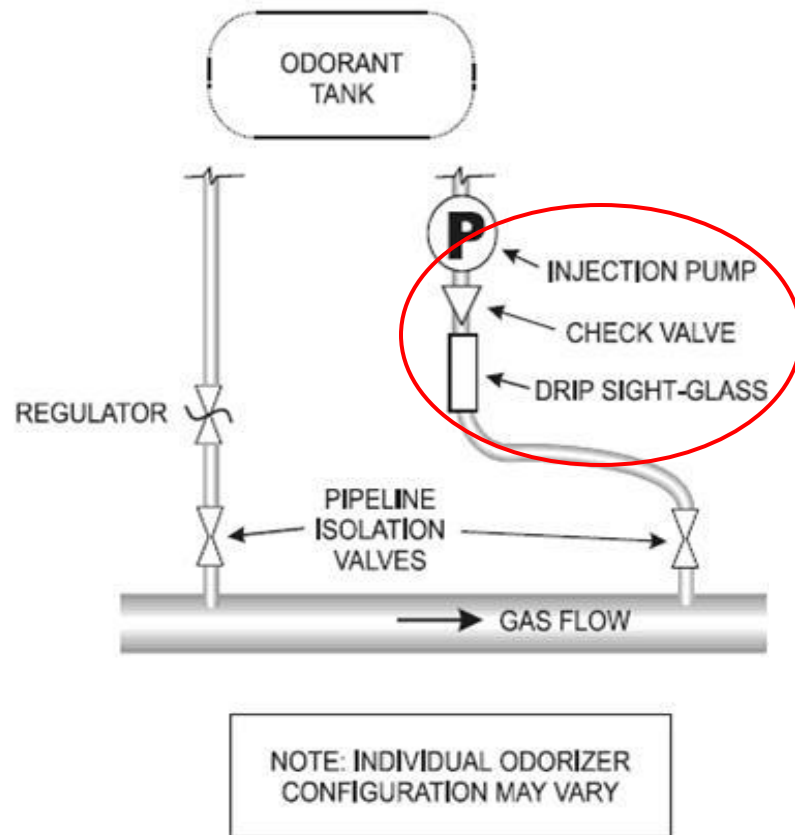
Bypass/ mini-bypass odorizers



Drip and injection odorizers

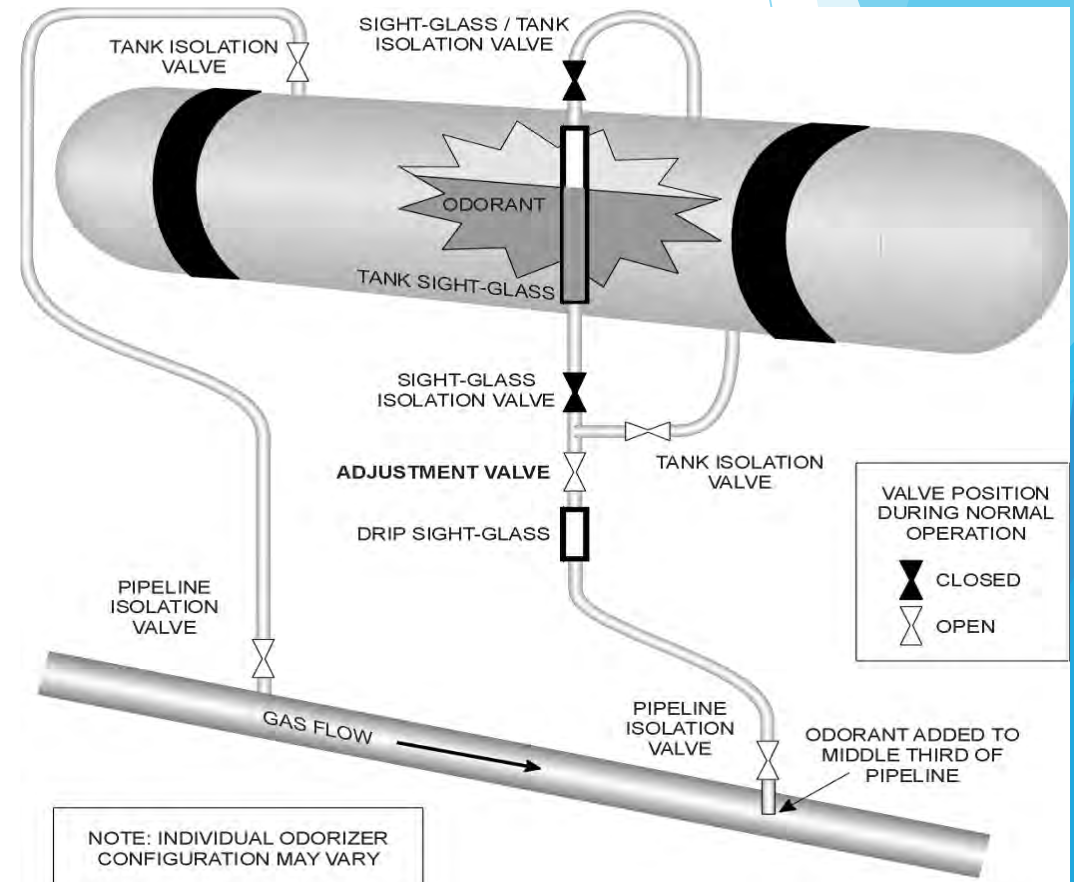
INJECTION

- ▶ Computer monitors the flow and adjusts the drip rate automatically
- ▶ More accurate



DRIP

- ▶ Manual drip rate change (Labour intensive)
- ▶ Difficult to adjust accurately



Injection odorizer



- ▶ Monitored by SCADA type systems for cc/stroke and flow rates
- ▶ Large system option, larger volume odorant tanks on site
- ▶ Typical rates of injection are 1lb/mmcf which is equal to 16mg/m³ Total odorant (TBM/MES)

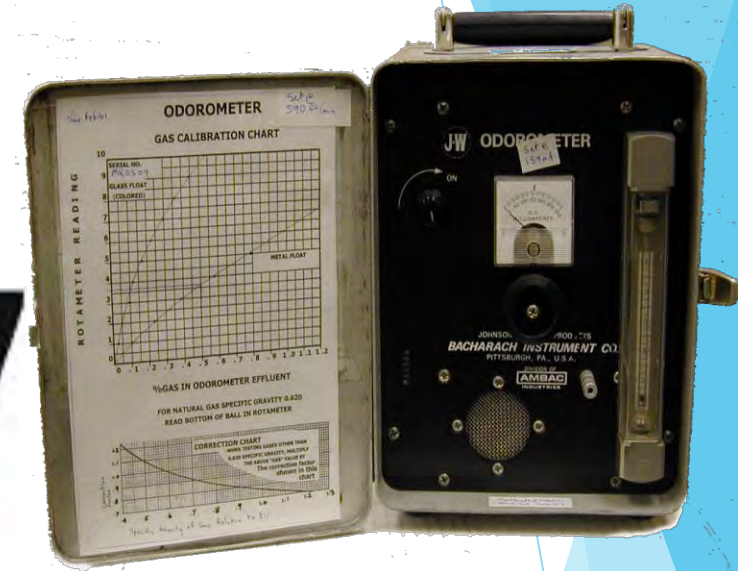
Low Volume Injection odorizer



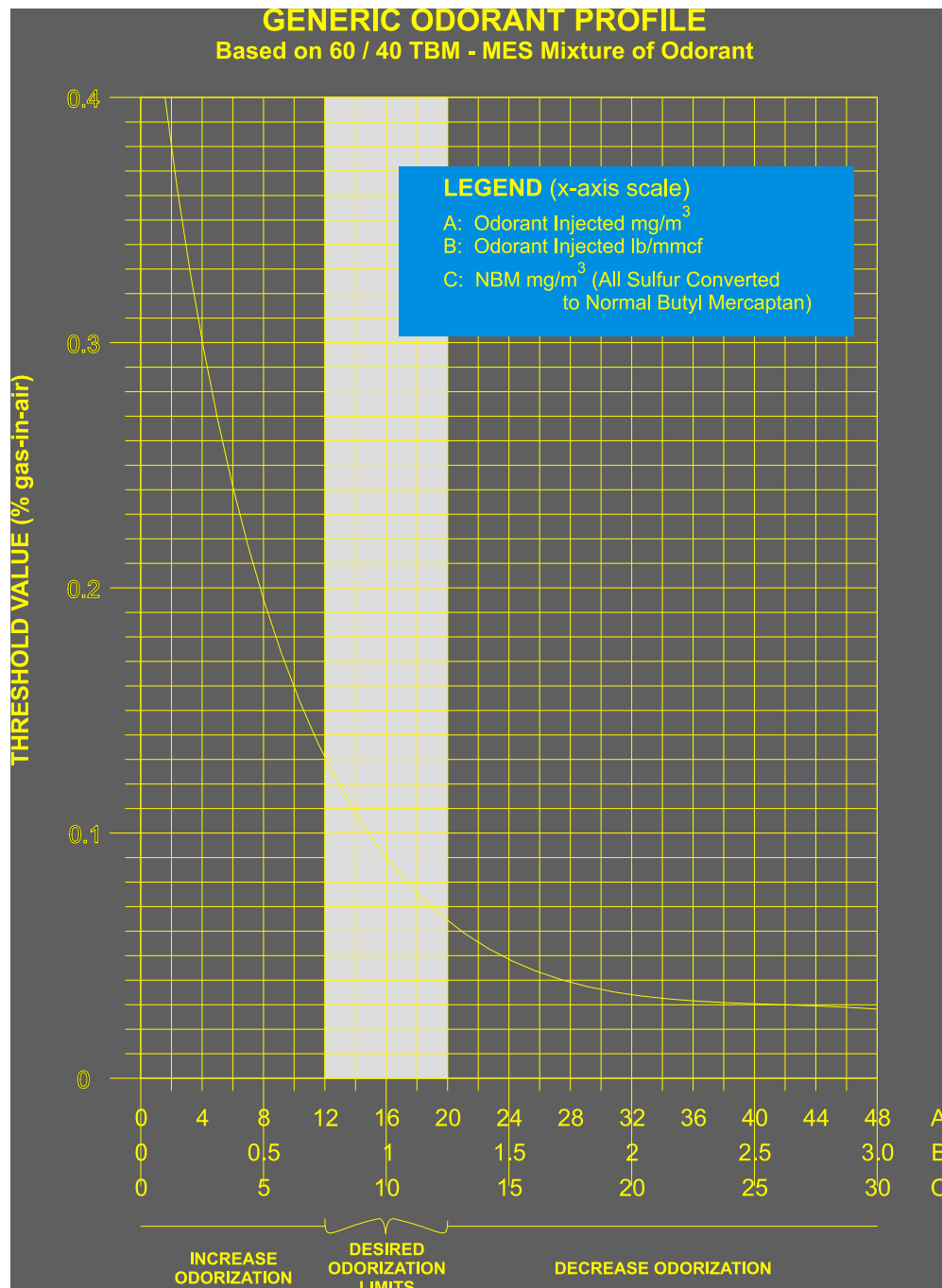
How do we ensure odorant thresholds are met

Odorometers

- ▶ These units provide a known mixture of Gas in Air (GIA) for testing, meeting the 1% GIA CSA Z662 requirement
- ▶ Units also provide safe test environment for employees only exposing them to 1% GIA



Nose profiles



Odorant Tank Level (cm): _____ Odorometer Serial #: _____ Resample ☐

Print Name & Sign	Threshold Detection Level % Gas in Air	Readily Detectable Level % Gas in Air	Odorant Intensity @ 0.5% Gas in Air	Odorant Odor Intensity Legend
				0 - Absent
				1 - Faint
				2 - Readily detectable
				3 - Strong
				4 - Very Strong

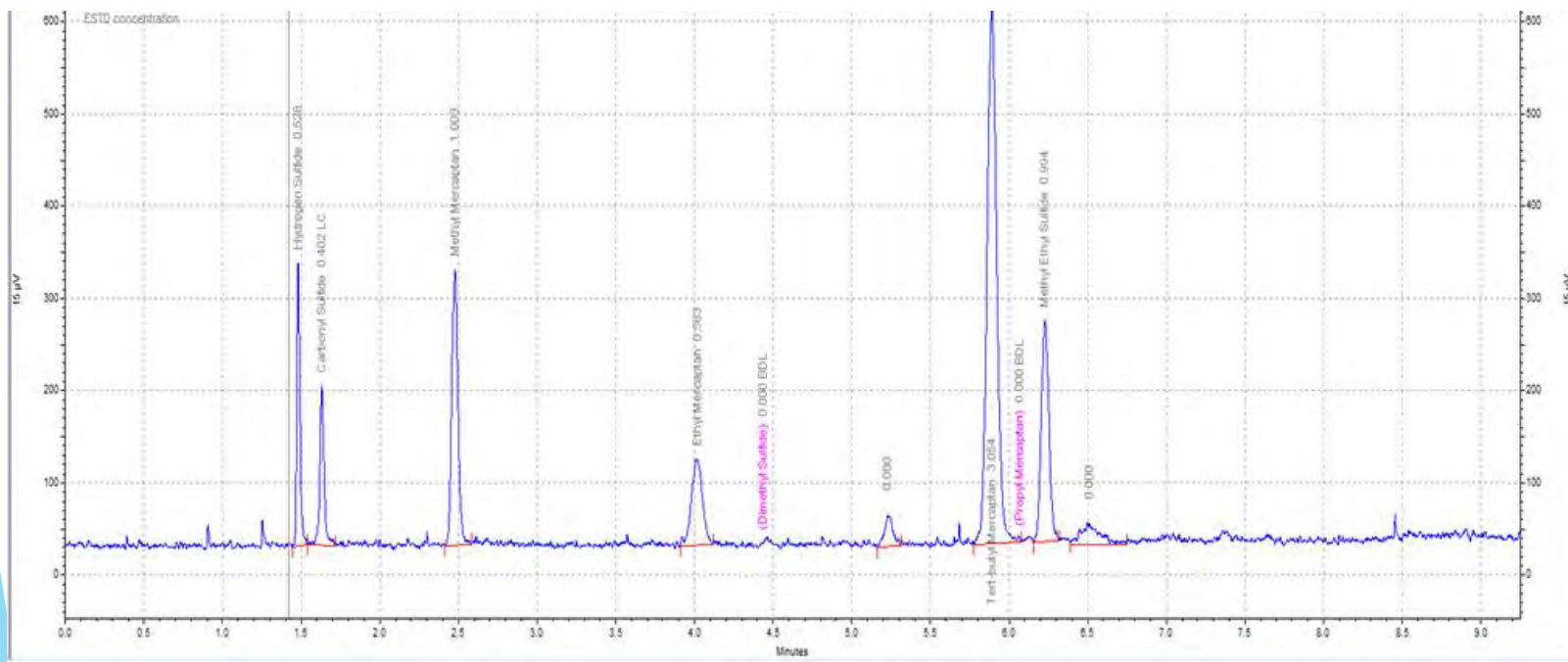
Comments: _____

0115-0038-0214

Laboratory analysis



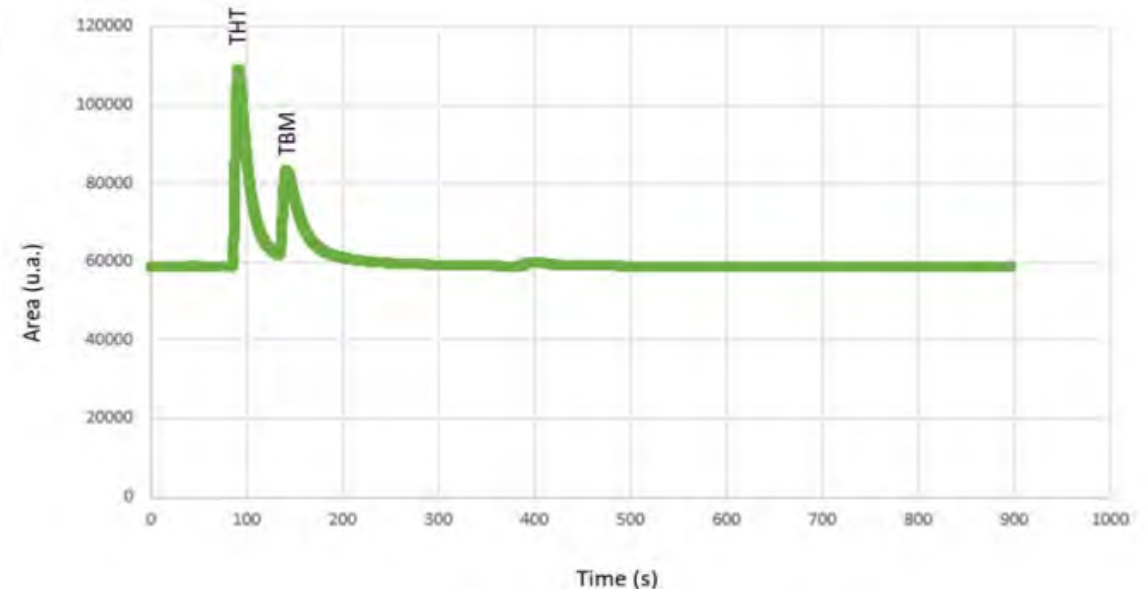
Analysis is done using a Gas chromatograph (GC) equipped with a sulfur Chemiluminescence detector (SCD) or flame photometric detector (FPD).



Online and hand-held devices

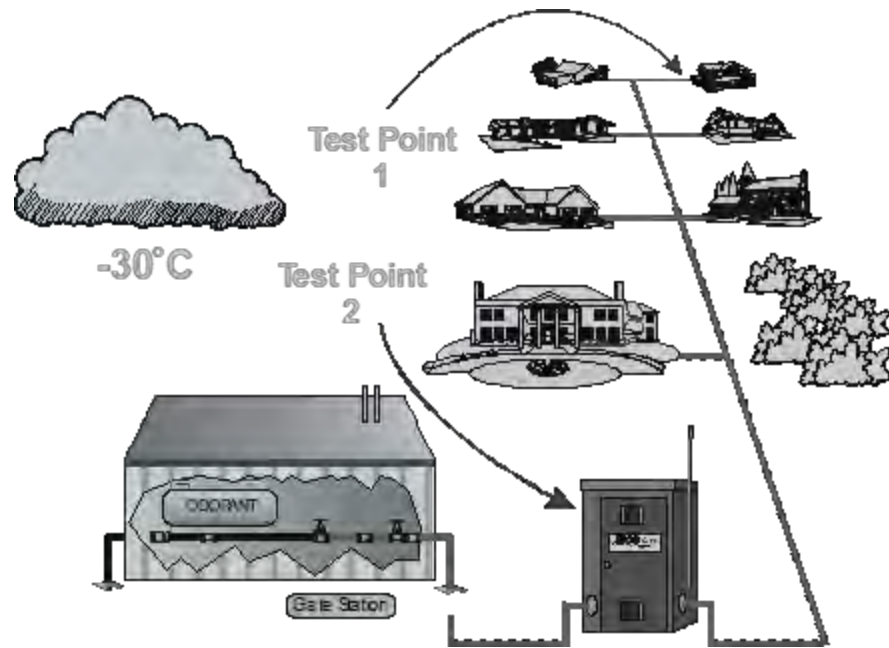


Compounds	Permeation rate	External permeation oven T°C	Concentration
THT	311 ng/min	60 °C	0.778 ppm
TBM	424 ng/min	60 °C	1.04 ppm



Where do we sample?

# of Customers	Min # of Sample Point Locations	Odorizer Sampling Frequencies
1 to 500	1	3-6 months (\pm 2 weeks- depending on reliability of odorizer)
501 to 5,000	2	2 months (\pm 2 weeks)
5,001 to 20,000	3	1 month (\pm 1 week)
> 20,000	4	2 weeks (\pm 1 week)



- ▶ Ideally near the end of the scheme
- ▶ Every odorizer in a system needs to be monitored to ensure customer safety
- ▶ More flow and more customers = more testing to minimize risk

Working with odorant

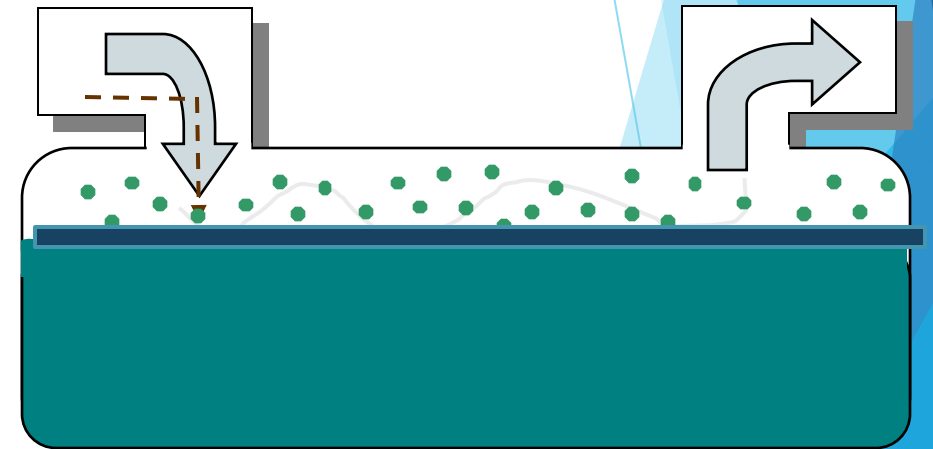
- ▶ Don't wear clothes you like!!!!
- ▶ Odorant has an affinity for metals, leather, hair.
- ▶ Wear your PPE
 - ▶ Gloves
 - ▶ Safety glasses
 - ▶ RPE (respiratory protective equipment)
 - ▶ Grounded Safety footwear
 - ▶ FR clothing



Constraints/ Interferences within Mercaptan Odorant Systems

Odorizer constraints

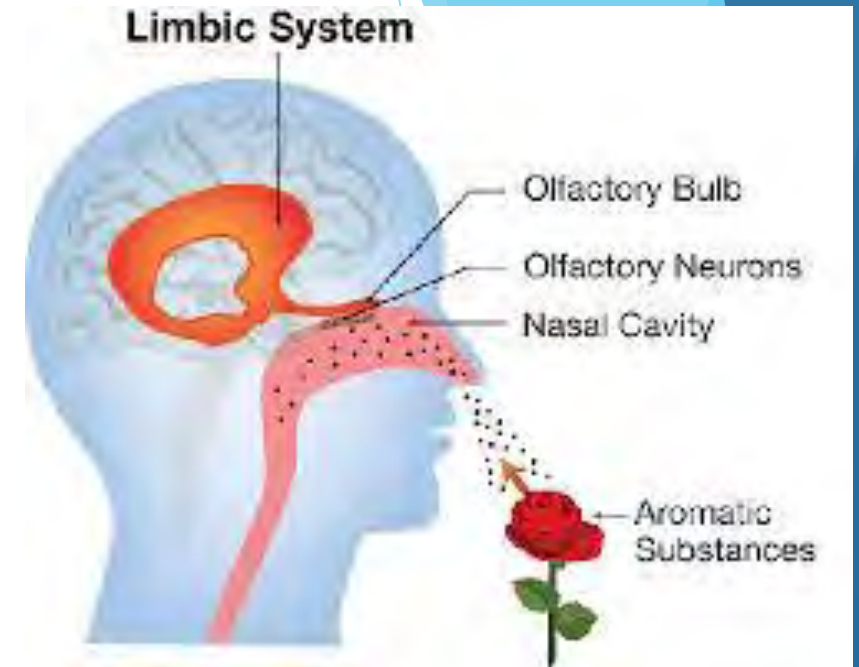
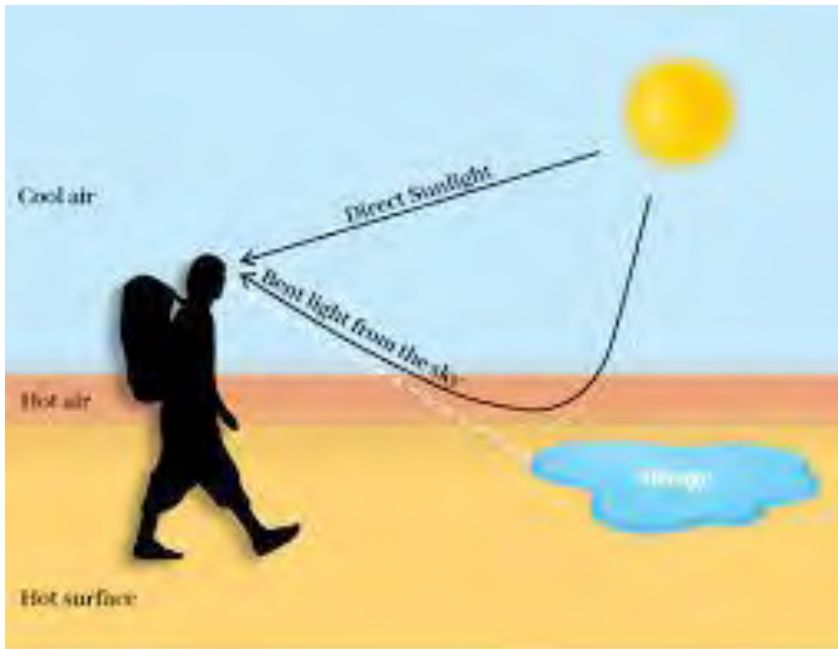
- ▶ Depending on selected odorant blend will determine acceptable odorizer choices
- ▶ For bypass odorizers:
 - ▶ Hydrocarbons can collect in the pot, creating an oily layer on top of the odorant which will inhibit the gas from interacting with the odorant vapour



Human Interferences

► Olfactory Fatigue

“is the temporary, normal inability to distinguish a particular odor after a prolonged exposure to that airborne compound” (*Odors chapter, Fundamentals volume of the ASHRAE Handbook, ASHRAE, Inc., Atlanta, GA, 2005*)

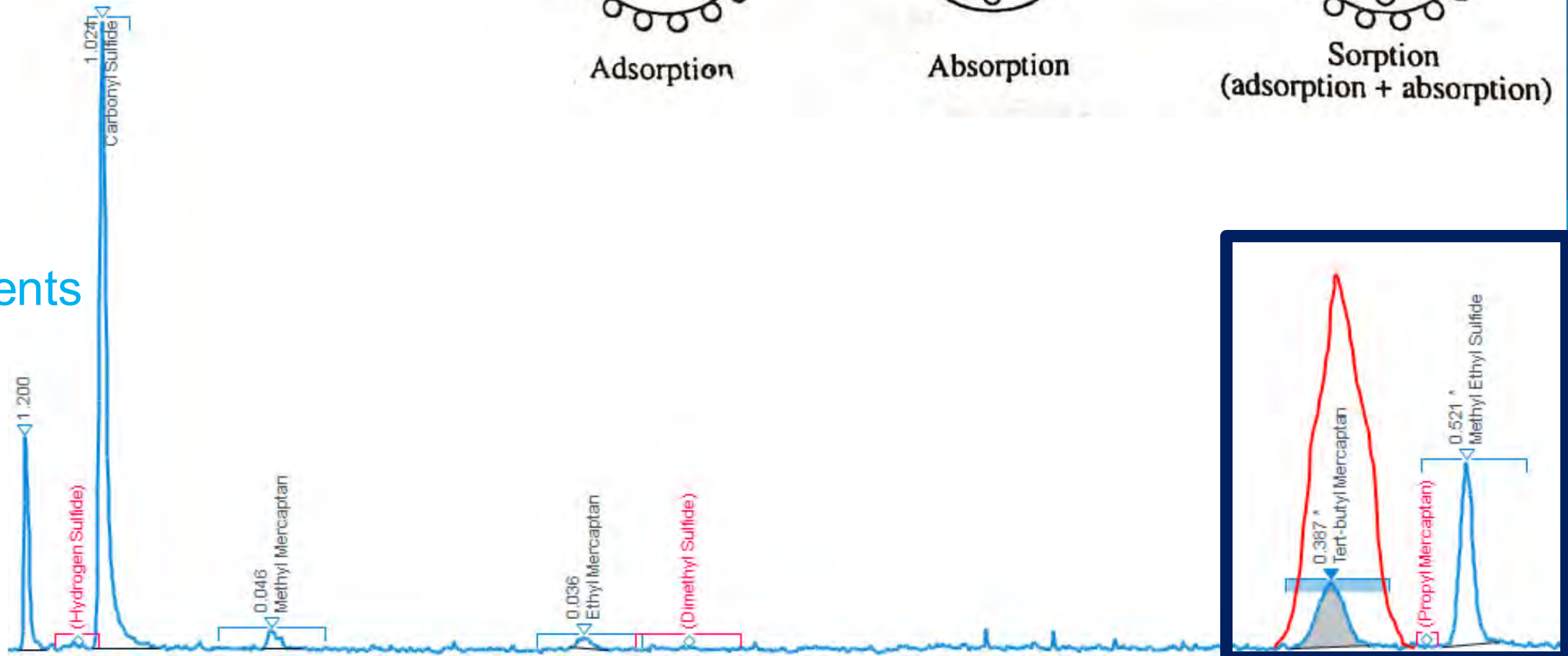
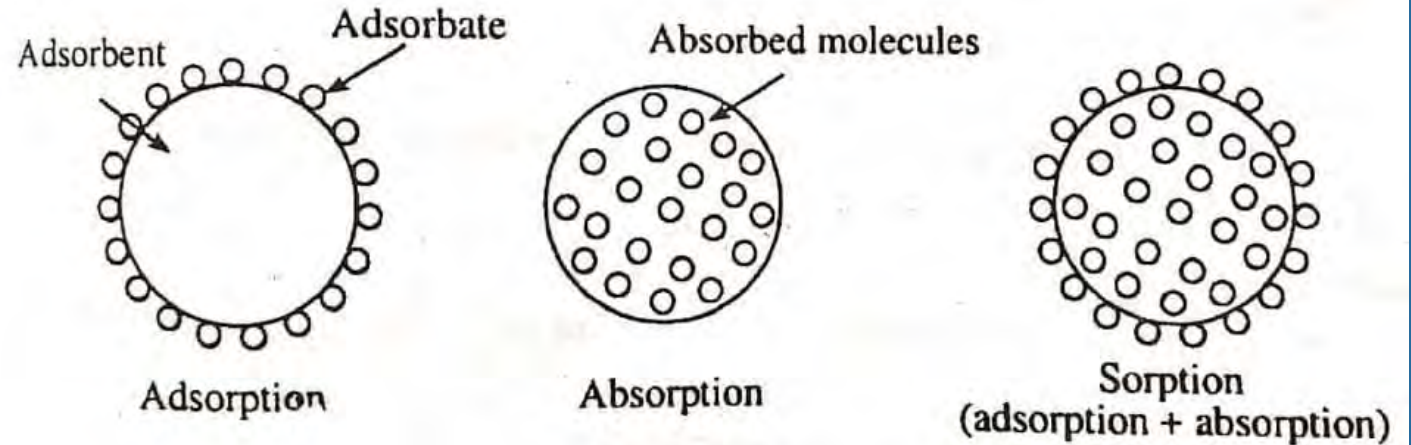


► Olfactory Mirage

Perception of an odor that isn't really there. Routine jobs with odor always present, could be a cause of this condition. (*AGA odorization manual 2017*)

Odorant interference

- ▶ Adsorption (Sticking Together)
- ▶ Absorption (Pulling Apart)
- ▶ Oxidation
- ▶ Masking Agents





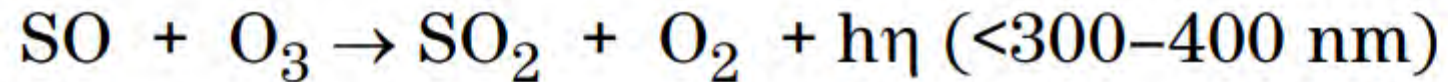
QUESTIONS??

References

- ▶ AGA odorization manual 2017
- ▶ www.arkema.com
- ▶ <https://naturalgasodorization.com>
- ▶ www.cpchem.com
- ▶ CSA Z662-15
- ▶ ATCO odorant monitoring course training material
- ▶ Arkema odorization presentation
- ▶ https://www.gasdetectorsusa.com/gdusa/download/Natural_gas_odorization.pdf
- ▶ file:///C:/Users/ymvj/Downloads/tsp_m41_m42_energyMEDOR.pdf
- ▶ 2022 Natural gas odorization conference presentations
- ▶ [SCD Manual.pdf](#)

How does the SCD and FPD detectors work?

Sulfur compound (analyte) \rightarrow SO + H₂O + other products



Stinky Stories

- ▶ Close to 10 years ago Contractor delivering odorant to Southern Alberta was stopped just prior to reaching USA border to return to Texas. The reason for the stop was Odor Calls spanning from Calgary to the Border that matched the route the driver was expected to take. Filling hose valve was not shut properly dripping residual odorant along the highway until driver was stopped to correct the leak.
- ▶ Chronic odor calls in Edmonton region- primarily around the time that odorant was delivered to brand new storage facility. After months of not finding the leak it was discovered that a bolt holding the inspection plate onto the tank had been bent when installing a walkway in the station and the smallest of pressure was leaking from that bolt causing the odor calls when the pressure was highest after filling.